

MASTERARBEIT | MASTER'S THESIS

Titel | Title

Pharmacological characterization of new psychoactive
substances

verfasst von | submitted by

Ronald Derndorfer BSc

angestrebter akademischer Grad | in partial fulfilment of the requirements for the degree of

Magister pharmaciae (Mag. pharm.)

Wien | Vienna, 2026

Studienkennzahl lt. Studienblatt |
Degree programme code as it appears on the
student record sheet:

UA 066 605

Studienrichtung lt. Studienblatt | Degree
programme as it appears on the student
record sheet:

Masterstudium Pharmazie

Betreut von | Supervisor:

ao. Univ.-Prof. Dr. Harald Sitte

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Vienna, May 2026

Ronald Derndorfer

Acknowledgement

I would like to thank my supervisor, Univ.-Prof. Dr. Harald Sitte, and my lab mentor, MSc. Nina Kastner, for their continuous support, as well as for their valuable explanations and advice throughout this master's thesis. They guided and supported me throughout the entire process. Furthermore, I would like to thank Julia Bicher, Marion Holy and Fatemeh Kooti for their support in the laboratory, particularly regarding practical work and training in good laboratory practice. I also thank all other members of the laboratory for their assistance.

Special thanks go to my parents, my sister, my uncle Peter, and my aunt Regina, who supported me in many different ways along my path. In addition, I would like to thank my close friends Lukas, Evelyn, Philipp, Bernhard, Julian, and Biko, who stood by my side at different stages of my life and inspired me. Thank you all. It has been an exciting journey so far, and I look forward to what lies ahead for all of us!

Ronald

Abstract

New psychoactive substances have gained increasing attention in public health and may also provide therapeutic options for psychiatric and cognitive disorders. MDMA is currently in Phase III clinical trials as an adjunct to psychotherapy for the treatment of PTSD. Despite two Phase III studies, the FDA issued a Complete Response Letter requesting additional studies to further evaluate safety and efficacy. This has driven the development of MDMA analogues with potentially improved pharmacological profiles.

In this master's thesis, six benzofuran cathinone derivatives were investigated *in vitro* for their activity at the monoamine transporters SERT, DAT, and NET using uptake inhibition and superfusion release assays in HEK-293 cells. IC₅₀ values and release properties were determined. The results indicated that stereochemistry, the position of the substituent on the benzofuran and the alkyl-chain length influence IC₅₀ values. Potency varied among the transporters with the highest potency at DAT, intermediate at NET, and the lowest at SERT. Release experiments showed that all substances induced release at SERT. At DAT, only *R*-βK-6-MAPB showed release, whereas at NET, both stereoisomers of βK-6-MAPB exhibited release.

These findings indicate that some of the tested benzofuran cathinones may offer selective serotonergic activity, highlighting their potential as candidates for further pharmacological evaluation for PTSD.

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List of Abbreviations

[³ H]-5-HT	Tritium-labelled 5-hydroxytryptamine (serotonin)
[³ H]-MPP ⁺	Tritium-labelled 1-methyl-4-phenylpyridinium
ADHD	Attention Deficit/Hyperactivity Disorder
S-AMPH	S-(+)-amphetamine
DMEM	Dulbecco's Modified Eagle's Medium
DAT	Dopamine transporter
DOx	4-substituted-2,5-dimethoxyamphetamines
FBS	Fetal bovine serum
HEK-293	Human Embryonic Kidney 293 cells
hNET	Human norepinephrine transporter
KHB	Krebs-HEPES buffer
LDX	Lisdexamfetamine
LSD	Lysergic acid diethylamide
MAO	Monoamine oxidase
MAT	Monoamine transporter
MDA	3,4-methylenedioxyamphetamine
MDMA	3,4-methylenedioxymethamphetamine
METH	Methamphetamine
MON	Monensin
NPS	New psychoactive substances
pCA	Para-chloroamphetamine
PDL	Poly-D-lysine
PTSD	Post-traumatic stress disorder
SERT	Serotonin transporter
SLC6	Solute carrier family 6
SOI	Substance of interest
SSRI	Selective serotonin reuptake inhibitor
TAAR1	Trace amine-associated receptor 1
VMAT	Vesicular monoamine transporter

1 Introduction

1.1 Monoamine neurotransmitters and their transporters

Monoaminergic neurotransmitters such as serotonin, dopamine, and noradrenaline are involved in neuronal signaling in the human brain (1). Their extracellular levels are primarily regulated by monoamine transporters (MATs), which belong to the solute carrier family 6 (SLC6). This transporter family includes the norepinephrine transporter (NET/SLC6A2), dopamine transporter (DAT/SLC6A3), and serotonin transporter (SERT/SLC6A4; 2,3). MATs are membrane-expressed proteins located at presynaptic terminals in the perisynaptic region (4), where they mediate the reuptake of neurotransmitters from the synaptic cleft following synaptic transmission (5). By removing released neurotransmitters from the extracellular space into neurons and glial cells, MATs contribute to neurotransmitter homeostasis and modulate both the duration and intensity of monoaminergic signaling (1).

The crucial physiological role of MATs is highlighted by their association with various neuropsychiatric and neurological disorders. Serotonergic signaling is closely linked to mood regulation, and altered SERT function has been implicated in depression and anxiety disorders (6,7). Dopaminergic signaling contributes to motivation, movement, mood, and learning, while dysfunction of DAT is associated with ADHD, Parkinson's disease, and substance abuse (8,9). Noradrenergic signaling is involved in cognition and emotional processing, and its dysregulation is associated with cognitive and affective disorders (10). Therefore, MATs are prominent pharmacological targets for antidepressants, psychostimulants, and various neuroactive drugs such as cocaine and amphetamines (1).

Structurally, all MATs share a conserved architecture composed of 12 α -helical transmembrane domains connected by intracellular and extracellular loops. Both the N- and C-termini are located on the intracellular side of the membrane. The transporters contain a primary high-affinity substrate-binding site (S1), where one substrate molecule binds together with one or two Na^+ ions. Within the S1 site, a hydrophobic region interacts with the aromatic moiety of the substrate, while a hydrophilic region contains a conserved aspartate residue that facilitates ionic interaction with the protonated amino group of the monoamine (5). In addition to the primary S1 site, a second binding site (S2) has been identified in SERT and NET (10). These structural features enable the conformational transitions that drive substrate transport.

Transport by MATs follows the alternating access model, in which conformational changes alternately expose the central binding site to either the extracellular or intracellular side of the membrane. In the outward-facing conformation, the S1 site is accessible to substrate and Na^+

binding. Substrate binding then triggers conformational transitions that lead to closure of the extracellular gate. Subsequently, the intracellular gate opens, resulting in an inward-facing conformation that enables release of the substrate into the cell. Hydration of the substrate-binding site facilitates this transition. After uptake, monoamines are either repackaged into synaptic vesicles for subsequent release or metabolized by monoamine oxidase (MAO) (5). Monoamine uptake and release by MATs are further influenced by the ionic environment surrounding the neurons. Monoamine uptake occurs through secondary active transport coupled to Na^+ and Cl^- . All MATs contain two conserved Na^+ binding sites (Na1 and Na2) and one conserved Cl^- binding site (10). Extracellular Na^+ , present at higher concentrations outside the cell, stabilizes the outward-facing conformation of the transporter (11). The Na^+ gradient that drives substrate uptake is maintained by the Na^+/K^+ -ATPase. Since monoamines are transported in their protonated form, substrate binding and transport strongly depend on electrostatic interactions (2,12). In addition to Na^+ and Cl^- ions, transporter activity is further influenced by intracellular K^+ . The return of SERT from the inward-facing to the outward-facing state constitutes the rate-limiting step of the transport cycle and is facilitated by cytoplasmic K^+ (6). In the presence of a substrate, SERT predominantly adopts the inward-facing conformation, as the transition from the inward- to the outward-facing state is slower than the reverse transition (6,13), whereas in the absence of substrate, SERT favors the outward-facing conformation (14). Under certain conditions, intracellular H^+ can substitute for K^+ , particularly in the presence of a transmembrane pH gradient that promotes uptake (6). Although a distinct K^+ binding site has not been conclusively identified (10), recent studies indicate that DAT can also use K^+ as a catalytic ion, while it is not essential for transport. Likewise, NET can also interact with K^+ at the Na1 site under certain conditions (10). Taken together, these observations underline the dynamic ion dependence of MAT-mediated transport.

Transporter function is regulated not only by ions but also by the lipid composition of the plasma membrane. Cholesterol, a key component of the membrane, contributes to modulating the conformational dynamics of MATs. Cholesterol depletion shifts MATs toward an inward-facing conformation, as shown for DAT (15–17) and SERT (18,19). Beyond these conformational effects, MATs can form higher-order oligomeric structures within the membrane, which contribute to the regulation and quality control of transporter function. In this context, altered cholesterol levels have been associated with several psychiatric disorders (20). Accordingly, MATs play a central role in regulating key monoamines in neuronal signaling. While MATs regulate extracellular monoamine levels through reuptake at the plasma membrane, intracellular monoamine storage depends on vesicular monoamine transporters

(VMATs). VMATs are membrane proteins located in synaptic vesicles that mediate the sequestration of monoaminergic neurotransmitters. In neurons, VMAT2 (SLC18A2) is the predominant isoform (21–23). VMAT2 operates as a proton-monoamine antiporter that exchanges two protons (H^+) from the vesicular lumen into the cytosol for one cationic monoamine (24). The proton gradient driving this process is generated by vesicular V-type H^+ -ATPases (25). By regulating intracellular monoamine storage and availability, VMAT2 plays a key role in neurotransmission. Amphetamines act as substrates of VMAT2, and this interaction substantially contributes to their overall effect on monoamine release (2). Alterations in VMAT2 activity have been associated with the neurotoxic effects of amphetamines, particularly through dysregulated dopamine handling (26).

1.2 Amphetamines

Amphetamines can be defined from different perspectives. From a chemical standpoint, Biel and Bopp defined amphetamines as compounds characterized by four structural features: (i) an unsubstituted phenyl ring, (ii) a two-carbon side chain linking the phenyl ring to a nitrogen atom, (iii) an α -methyl group, and (iv) a primary amino group (27).

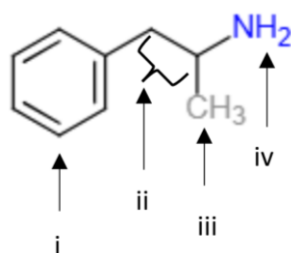


Figure 1: Structural definition of amphetamines according to Biel and Bopp: (i) an unsubstituted phenyl ring, (ii) a two-carbon side chain between the phenyl ring and nitrogen, (iii) an α -methyl group, and (iv) a primary amino group (27).

The term “amphetamine” is derived from α -methylphenethylamine (28). However, this chemical definition is relatively strict, as several amphetamine-related compounds, including methamphetamine (METH) and ephedrine, do not fulfill the primary amine requirement despite sharing similar pharmacological properties (2). Conversely, a purely operational definition based on molecular targets is also insufficient, since amphetamines can exert diverse effects, including inhibition of neurotransmitter reuptake and induction of neurotransmitter release (2). The primary molecular targets of amphetamines are MATs (29). Unsubstituted amphetamines generally exhibit higher potency at catecholamine transporters, particularly NET, while displaying lower potency at SERT (30). Structural modifications, especially substitutions on the phenyl ring, strongly influence transporter selectivity and pharmacological activity. Ring substitutions typically increase binding affinity for SERT (31). For example, *para*-chloroamphetamine (*p*CA) and fenfluramine display greater selectivity for SERT than DAT, whereas *S*-(+)-amphetamine (*S*-AMPH) preferentially targets DAT. Amphetamines enter monoaminergic neurons either through passive diffusion across the plasma membrane or via transporters such as SERT, DAT, and NET (32). Once inside the cell, they can subsequently enter synaptic vesicles through VMAT2-mediated transport (28). Since VMAT2 activity depends on the acidic vesicular environment, vesicular uptake of amphetamines is strongly influenced by proton gradients (28). This process represents a key step in the mechanism of action of amphetamines and forms the basis of the weak base hypothesis.

The weak base hypothesis proposes that amphetamines, which are weak bases ($pK_a \approx 8$ -10), can passively diffuse across the vesicular membrane in their neutral form, independently of

transporter-mediated uptake. Once inside the vesicle, they become protonated, leading to vesicular alkalization and reduced membrane permeability. As a consequence, VMAT function becomes impaired, promoting the accumulation of monoamines within the cytosol (28). However, this hypothesis alone cannot fully explain amphetamine action, since stereospecific differences have been observed. For instance, *S*-(+)-METH induces vesicular monoamine release more effectively than *R*-(-)-METH. Therefore, passive diffusion is best regarded as complementary to the classical substrate-mediated VMAT transport (33). In addition to their effect on MATs and VMATs, amphetamines also affect intracellular monoamine metabolism. Amphetamines are not primarily metabolized by monoamine oxidase (MAO) (28,34), but they can inhibit MAO activity, thereby reducing monoamine degradation and increasing cytosolic monoamine levels (2). In addition, MAO-A exhibits stereoselectivity toward AMPH, with higher affinity for the *S*-enantiomer than for the *R*-enantiomer (35,36). However, the affinity of amphetamines for MAO is substantially lower than for MATs (28). The resulting increase in cytosolic monoamine availability contributes to reverse monoamine transport mediated by MATs.

Several mechanisms have been proposed to explain transporter-mediated reverse transport. Amphetamines increase the rate of outward-to-inward transporter translocation, thereby increasing the likelihood that intracellular dopamine binds to the transporter and is subsequently released into the extracellular space (28). In addition, reverse transport may occur through a channel-like transporter state, in which substrate efflux proceeds in an uncoupled manner, resembling ion conductance through classical ion channels (28). Intracellular Na⁺ concentrations further increase as a consequence of AMPH-induced transport activity, which additionally promotes reverse monoamine transport via DAT (37). Second messenger signaling pathways also contribute to AMPH-mediated dopamine efflux (4). Apart from these mechanistic aspects, structural modifications of amphetamines also strongly influence transporter selectivity and releasing properties.

Many amphetamines act as more potent releasers at NET than at DAT (39). In contrast, substitutions on the phenyl ring of amphetamines enhance serotonin release via SERT (31,40). Increasing N-alkyl chain length in amphetamines, with the exception of AMPH and METH, reduces binding potency at MATs while exerting comparatively minor effects on transport inhibition (41,42). Beyond transporter-mediated effects, amphetamines also interact with additional receptor and signaling systems. In particular, amphetamines act as agonists at trace amine-associated receptor 1 (TAAR1) (43,44). Furthermore, amphetamines can activate presynaptic α 2-adrenergic receptors (45) and serotonin receptors of the 5-HT₂ family.

Para-substituted amphetamines preferentially activate 5-HT_{2A} receptors (31), whereas ring-substituted amphetamines can activate 5-HT_{2A}, 5-HT_{2B}, and 5-HT_{2C} receptors (46,47).

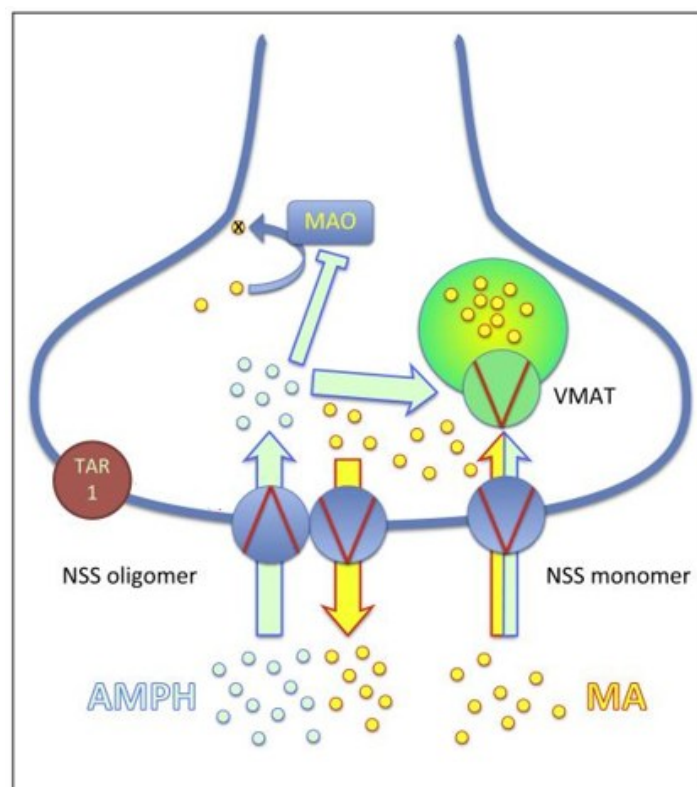


Figure 2: Mechanistic insights into amphetamine action. Adapted from Sitte & Freissmuth, *Trends Pharmacol Sci.* 2014; 36(1):41–50. doi: 10.1016/j.tips.2014.11.006

1.2.1 Amphetamine analogues

1.2.1.1 Classical (unsubstituted) amphetamines

Amphetamines are among the most commonly abused illicit substances (48). Classical amphetamines include AMPH and METH. Their pharmacological effects are primarily mediated by the release of catecholamines. Acute effects include euphoria and increased stimulation, whereas chronic or high-dose use is associated with anxiety, psychosis, and addiction (48,49).

1.2.1.2 Substituted amphetamines

Para-substituted amphetamines exhibit a more pronounced serotonergic profile than unsubstituted amphetamines and show increased affinity for the 5-HT_{2A} receptors (31). A distinct subgroup comprises the DOx compounds (4-substituted-2,5-dimethoxyamphetamines), whose effects are primarily mediated through 5-HT_{2A} receptor agonism rather than MAT interactions. Consequently, these compounds are classified as psychedelic substances with effects resembling those of classical psychedelics such as lysergic acid diethylamide (LSD) or mescaline (50).

Ring-substituted amphetamines such as MDMA and 3,4-methylenedioxyamphetamine (MDA) act as releasers at all MATs. Compared to MDMA, MDA exhibits a higher DAT/SERT ratio and stronger 5-HT_{2A} agonism, which may contribute to more pronounced stimulant effects (51). MDMA is commonly described as an empathogen and entactogen, reflecting its ability to enhance prosocial behaviour, empathy, openness, and self-awareness (52). Consistent with these behavioural effects, MDMA alters emotional processing, including reduced recognition accuracy of negative emotions in women as well as enhanced emotional empathy toward positive stimuli and prosocial behaviour in men (53). These characteristic behavioural effects of MDMA are closely linked to its pharmacological actions at different targets.

MDMA acts as a substrate of all MATs, but with a particular emphasis of the SERT. MDMA induces transporter reversal at all MATs (40,54,55). At SERT, MDMA functions as a full releaser (56), whereas at DAT, MDMA induces only partial monoamine release (56). MDMA also acts as a substrate at NET, inducing transporter reversal (54). SERT-mediated serotonin release is considered a key contributor to MDMA's psychological (57) and physiological (e.g., cardiovascular) responses (58). MDMA-induced transporter reversal at NET contributes to sympathetic activation (54). This NET-mediated mechanism is thought to underlie some of the cardiovascular and psychotropic effects of MDMA (59). Overall, MDMA induces measurable monoamine release at all three MATs, with effect sizes following the order SERT > NET > DAT (56). Ring substitution reduces the DAT:SERT inhibition ratio, with 3,4-substitution causing a particularly pronounced decrease. Compared with other amphetamines, only a limited number of compounds induce release at both serotonergic and catecholaminergic systems (42,60). MDMA displays a DAT:SERT release ratio < 1 (60), and in some cases as low as 0.1 (56). Compounds with relatively higher serotonin release compared to dopamine are associated with reduced reinforcing effects in self-administration models (61). Consistent with this observation, activation of 5-HT_{2C} receptors within the nucleus accumbens appears to contribute to the comparatively low abuse potential of MDMA (62). Beyond its influence on reinforcement, the pronounced serotonergic profile of MDMA is also thought to underlie its characteristic prosocial and empathogenic properties, possibly through oxytocinergic signaling pathways (63,64). Oxytocin, a neuropeptide involved in emotion recognition, prosocial behaviour (65), and altruism (66), has been implicated in several behavioural effects of MDMA. In humans, MDMA increases peripheral oxytocin concentrations (67), an effect that is likely mediated, at least in part, via 5-HT_{1A} receptor activation (63). Elevated oxytocin levels are associated with enhanced sociability and subjective friendliness (64). In addition, 5-HT_{1B} receptors also appear to contribute to MDMA-induced

prosocial effects (68). While serotonergic signaling contributes to both the reduced abuse potential and the prosocial effects of MDMA, interactions with adrenergic and pituitary pathways shape its physiological responses. α 1- and β -adrenergic receptors are involved in the thermogenic and cardiovascular effects of MDMA (69). Besides its pharmacodynamic profile, MDMA is also characterized by distinct pharmacokinetic properties. MDMA is metabolized into several active metabolites, including 3,4-methylenedioxyamphetamine (MDA). Interindividual variability in MDMA metabolism is influenced by genetic polymorphisms, particularly in CYP2D6 (70). Many amphetamines inhibit CYP2D6, an effect that is enhanced by the 3,4-methylenedioxy substitution (71). MDMA exhibits a broad range of side effects, the severity of which depends on factors beyond dose and route of administration (72), including species (73,74), age (75,76), body temperature (77), sex (76), and genetic polymorphisms (78). Among the adverse effects, particular attention has been given to the potential neurotoxic effects of MDMA. Ring-substituted amphetamines induce degenerative changes in various rat brain regions (79). However, the doses at which potentially neurotoxic effects have been described (79) are often much higher than those that elicit relevant neurochemical, endocrine, and behavioural effects (80). A meta-analysis of 16 studies reported reduced SERT density in 8 out of 13 brain regions in human participants, many of whom were heavy MDMA users with concurrent polysubstance use. This reduction is commonly interpreted as a marker of serotonergic neurotoxicity (81). However, some studies suggest that decreases in SERT density may be at least partially reversible (82–84). While SERT density may recover after abstinence, memory-specific deficits do not appear to normalize (83). These serotonergic alterations may contribute to the cognitive deficits observed in human MDMA users. Heavy MDMA users appear to be at increased risk of impairments in executive updating, attention, and working memory (85). Regarding MDMA-related cognitive effects, cumulative lifetime exposure does not appear to be the primary determinant. Rather, the dose per session may be more critical (86). Several mechanisms have been proposed to explain MDMA-induced neurotoxicity, but their relative contributions remain unclear. It is likely that a combination of factors, including oxidative stress, toxic MDMA metabolites, mitochondrial dysfunction, glial cell involvement, excitatory events, and hyperthermia, acts synergistically to produce the observed neurotoxic effects (87).

1.2.1.3 Cathinones (β -keto amphetamines)

Cathinones, also known as β -keto amphetamines, are characterized by the presence of a β -keto group (88). They form the structural basis of many designer drugs that are marketed as “bath salts” or “plant food” to evade legal control. Examples include

mephedrone (4-methylcathinone), methylone (3,4-methylenedioxymethcathinone), and MDPV (3,4-methylenedioxypropylvalerone). Synthetic cathinones are among the most frequently seized new psychoactive substances (NPS). They resemble classical psychostimulants such as amphetamine, cocaine, and MDMA in their effects (89). Although all cathinones target MATs, their mechanisms of action vary considerably. They may act as releasers, uptake inhibitors, or exhibit mixed activity by functioning as a releaser at one transporter and as an inhibitor at another. Such compounds are commonly referred to as hybrid compounds (90).

1.2.1.4 Benzofuran derivatives

Aminoalkylbenzofurans and N-methylated aminoalkylbenzofurans are NPS that, similar to MDMA, act as substrates at MATs. All tested benzofurans exhibit relatively similar binding affinities for NET in the submicromolar range, suggesting psychostimulant and cardiostimulant effects comparable to those of MDMA. In contrast, their binding affinities for SERT and DAT vary considerably, resulting in DAT:SERT inhibition ratios ranging from 0.01 to 0.65. Compared with MDMA and other entactogens, which typically display ratios between 0.01 and 1, these findings suggest that benzofurans may also exhibit entactogenic effects (41). If the oxygen atom is located at the para position in benzofurans, serotonergic properties are enhanced relative to dopaminergic ones (41,91). Entactogenic effects have been reported for both para- and meta-substituted benzofurans (92–94). Although benzofurans have not been evaluated in self-administration models, compounds with higher serotonin release relative to dopamine typically show reduced reinforcing effects in such models (61). Accordingly, benzofurans, which exhibit lower DAT:SERT ratios, may have a reduced abuse potential (41). Beyond their effects on MATs, benzofurans may also interact with adrenergic and serotonergic receptors. Certain benzofurans show binding affinity for 5-HT_{2A} receptors ($K_i < 1\mu\text{M}$) and display moderate functional activity ($\text{EC}_{50} \approx 6\text{--}13\mu\text{M}$; $\text{E}_{\text{max}} \approx 20\text{--}60\%$). Compared with MDMA ($K_i = 6.3\mu\text{M}$; $\text{EC}_{50} = 6.1\mu\text{M}$; $\text{E}_{\text{max}}(\text{MDMA}) = 55\%$), they exhibit higher binding affinity, while functional potency and efficacy vary across compounds and are often comparable to or somewhat lower than those of MDMA (41). In contrast to MDMA, some benzofurans also exhibit a high binding affinity for 5-HT_{2C} receptors ($K_i \approx 0.1\mu\text{M}$) and functional potency at 5-HT_{2B} receptors ($\text{EC}_{50} \approx 0.1\text{--}0.3\mu\text{M}$) (41). Some benzofurans show high binding affinity for α_2 -adrenergic receptors ($K_i < 1\mu\text{M}$), which is associated with modulation of norepinephrine signaling. Interactions with α -adrenergic receptors contribute to sympathomimetic effects and may be more pronounced than those of MDMA (41). In addition to interactions with MATs and serotonergic and adrenergic receptors, TAAR1 binding

represents another mechanism through which benzofurans exert their effects. Nearly all benzofurans show strong binding affinity for TAAR1 ($K_i \approx 0.1\text{-}1\ \mu\text{M}$), which lies within a similar range as MDMA ($K_i = 0.37\ \mu\text{M}$). Cathinone derivatives show approximately tenfold lower affinity at TAAR1 compared with their non-keto analogues (41,42). TAAR1 is thought to mediate autoinhibitory mechanisms that limit locomotor and stimulant effects (41). These pharmacological differences may also translate into distinct toxicological and abuse-related profiles compared with MDMA.

Comparisons of structurally related compounds (e.g., MDMA vs. methyline, MBMD vs. butylone, and MDEA vs. ethylone) show that the DAT:SERT ratio increases by 6- to 41-fold, which may be associated with higher abuse liability (42). However, this stronger dopaminergic effect may be counteracted by more pronounced 5-HT_{2C} receptor activity. As benzofurans exhibit higher 5-HT_{2C} receptor binding, this may attenuate the reinforcing component (41). Furthermore, the increased 5-HT_{2B} receptor binding affinity of benzofurans compared with MDMA (41) may be associated with an elevated risk of valvular heart disease (VHD) (46,95). For benzofuran cathinones, reduced TAAR1 affinity compared with amphetamines may result in more pronounced stimulant effects and potentially higher abuse potential (41). Some cathinones (e.g., MDMA vs. methyline) show lower VMAT2 interaction, including reduced uptake and maximal release (96,97). Since dopamine release via VMAT2 has been associated with neurotoxic effects (26), benzofuran cathinones may represent promising candidates for further investigation compared with classical benzofuran amphetamines.

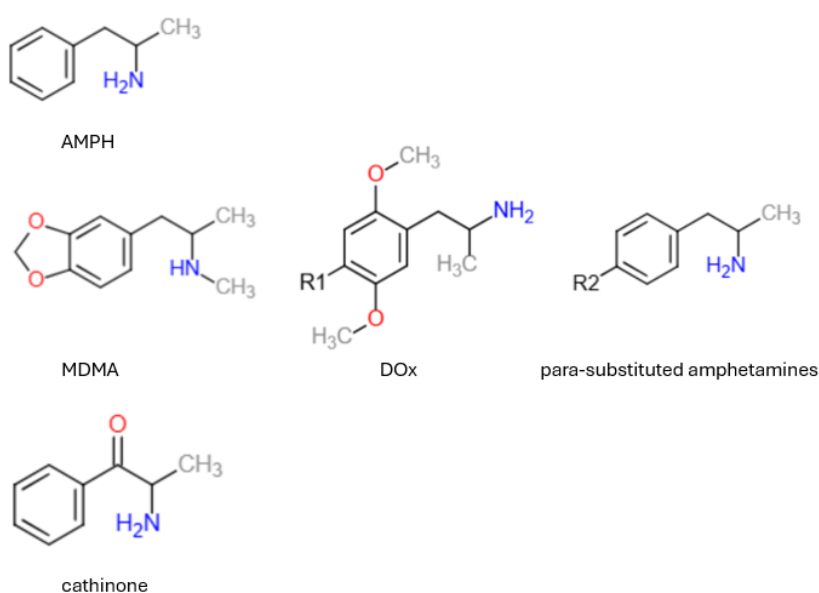


Figure 3: Chemical structures of selected subgroups of amphetamine derivatives.
In this case R₁ and R₂ represent halogens (F, Cl, Br and I)

1.2.2 Therapeutic use of amphetamines

Amphetamines have therapeutic applications primarily in the treatment of neuropsychiatric disorders. For example, several monoaminergic drugs are used in the treatment of attention-deficit/hyperactivity disorder (ADHD). These include methylphenidate, which acts primarily as a DAT inhibitor (5), Lisdexamfetamine (LDX), a prodrug of *S*-amphetamine that functions as a monoamine-releasing agent (98), and mixed amphetamine salts containing both *S*- and *R*-amphetamine (99). Both methylphenidate and amphetamine are also used to treat narcolepsy (100). Bupropion, a cathinone derivative, is approved as a norepinephrine-dopamine reuptake inhibitor (NDRI) for the treatment of depression (101,102) and smoking cessation (103). Tranylcypromine and selegiline are also classified as amphetamines and are approved for depression (104) and Parkinson's disease (105), respectively.

Finally, MDMA shows potential for therapeutic use in post-traumatic stress disorder (PTSD). Despite the availability of multiple treatment options, treatment resistance remains high, with a non-response rate of approximately 39% (106). This underscores the need for novel therapeutic approaches for PTSD, particularly in the pharmacological domain. In recent years, an increasing number of studies have reported beneficial effects of MDMA-assisted psychotherapy in patients with PTSD (107,108). The therapeutic effects of MDMA in PTSD may be mediated by enhanced fear extinction (109,110). Furthermore, the SERT and 5-HT_{2A} receptors appear to be involved in MDMA-enhanced fear extinction (109). While these findings support a role for specific receptors in fear extinction, results from animal models are not entirely consistent, highlighting the need for translation to human studies (110,111). In addition to its effects on fear extinction, MDMA may enhance its therapeutic impact in PTSD through its prosocial effects, which may facilitate patients' willingness to discuss traumatic experiences (112). These considerations have motivated the search for MDMA-like compounds with potentially improved pharmacological and toxicological profiles.

Benzofuran derivatives may differ from MDMA in their neurotoxic potential, likely due to differences in MAT and VMAT2 interactions. Therefore, this study investigates the effects of selected benzofuran cathinones on MATs (SERT, DAT, and NET), focusing on both uptake inhibition and release assays.

1.3 Introduction to cell lines and *in vitro* assays

For the experiments conducted in this thesis, HEK-293 cells were used due to their ease of transfection, rapid growth, and high reproducibility (113). These cells were used as a

heterologous expression system for MATs. To characterize the pharmacological properties of the investigated compounds, uptake inhibition and superfusion release assays were employed. Uptake inhibition assays measure transporter-mediated uptake of radiolabeled substrates into cells (114) and allow determination of the half-maximal inhibitory concentration (IC_{50}) (115). However, these assays do not distinguish between substrates (e.g., amphetamines) and non-transported competitive inhibitors (116,117). Consequently, amphetamines produce dose-dependent inhibition of uptake in these assays (117). To address this limitation, superfusion release assays were employed. These assays measure transporter-mediated substrate efflux and thereby allow differentiation between releasers and uptake inhibitors (118).

2 Materials and methods

2.1 Investigated compounds

Six compounds were examined in this study for their activity in uptake inhibition and superfusion release assays at MATs. All compounds are benzofurans. They differ in the position of the substituent on the benzofuran ring, indicated by the numbering in their names. One group carries the substituent at position 5, whereas in the other it is located at position 6. The length of the alkyl side chain was varied, with compounds predominantly containing either a propyl or a butyl group. This is reflected in the third letter of their names (P for propyl and B for butyl). All tested compounds also contain a carbonyl function in the β -position, denoted by β K (β -keto) in their names. The letter M indicates the presence of an N-methyl group, and A denotes the amino function.

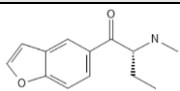
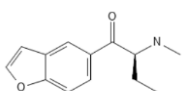
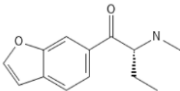
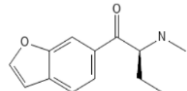
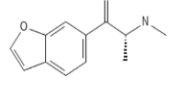
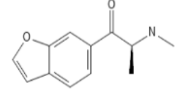
Substance	IUPAC name	Structure	M(as HCl) [g/mol]
<i>S</i> - β K-5-MABB	(<i>S</i>)-1-(benzofuran-5-yl)-2-(methylamino)butan-1-one		253.57
<i>R</i> - β K-5-MABB	(<i>R</i>)-1-(benzofuran-5-yl)-2-(methylamino)butan-1-one		253.57
<i>S</i> - β K-6-MABB	(<i>S</i>)-1-(benzofuran-6-yl)-2-(methylamino)butan-1-one		253.57
<i>R</i> - β K-6-MABB	(<i>R</i>)-1-(benzofuran-6-yl)-2-(methylamino)butan-1-one		253.57
<i>S</i> - β K-6-MAPB	(<i>S</i>)-1-(benzofuran-6-yl)-2-(methylamino)propan-1-one		239.55
<i>R</i> - β K-6-MAPB	(<i>R</i>)-1-(benzofuran-6-yl)-2-(methylamino)propan-1-one		239.55

Figure 4: Overview of the tested compounds, including their abbreviations, IUPAC names, structural formulas and molecular masses.

The enantiopure HCl salts of the benzofuran compounds in Figure 6 were generously supplied by Dr. Matthew J. Baggot (Tactogen Inc., Palo Alto, USA). Radiolabelled tracers [3 H]-5-HT (NET498001MCSBR1) and [3 H]-MPP⁺ (ART 0150) were obtained from Revvity and American Radiolabeled Chemicals Inc. (ARC), respectively. Paroxetine (cat. no. AB 439408)

was obtained as its hydrochloride form from aber GmbH (Karlsruhe, Germany). GBR12909 (as dihydrochloride) (cat. no. D052), *S*-(+)-AMPH (as its hemisulfate salt) (cat. no. A-5880), *p*CA (cat. no. C9635) as well as MON (as sodium salt) (cat. no. M5273) were received from Sigma-Aldrich (St. Louis, MO, USA). Ultima GoldTM XR (cat. no. 6013117) scintillation cocktail was provided by PerkinElmer (Waltham, USA) and used for 96-well-based assays. Rotiszint[®]Routine (1P1A.2) scintillation cocktail was obtained from Carl Roth (Karlsruhe, Germany) and used for release assays.

2.2 Cell culture

HEK-293 cells that stably express the human isoforms of SERT (= hSERT), DAT (= hDAT), and NET (= hNET) were used. Additionally, hSERT and hDAT were N-terminally tagged with YFP, whereas hNET was untagged. These stable cell lines were kindly provided (115). Cells were cultured in Dulbecco's Modified Eagle's Medium (DMEM) (cat. no. L0102) from Biowest (Bradenton, USA), which was supplemented with 10% fetal bovine serum (FBS) (cat. no. FBS-11A) from Capricorn Scientific GmbH (Ebsdorfergrund, Germany). Once the cells reached a density of approximately 80%, they were passaged to maintain a sub-confluent state or to prepare them for seeding in experimental plates (115). This was done by washing with 4-5 mL phosphate-buffered saline (PBS; 137 mM NaCl, 4.3 mM Na₂HPO₄·2 H₂O, 2.7 mM KCl, and 1.5 mM KH₂PO₄, adjusted with NaOH to pH 7.3). Afterwards, 1 mL of pre-warmed trypsin-EDTA solution (cat. no. T3924) was added to detach the cells. After 1 min, trypsinization was stopped by adding 9 mL of pre-warmed DMEM. The cells were then triturated to obtain a single-cell suspension (115). To ensure stable expression of MATs in HEK-293 cells, selection was maintained by adding 50 µg/mL Geneticin (G418) (cat. no. G418-B) from Capricorn Scientific GmbH (Ebsdorfergrund, Germany). Before seeding, 96-well plates were coated with poly-D-lysine (PDL) (as its hydrobromide form) (cat. no. P1149) from Sigma-Aldrich (St. Louis, MO, USA) the day before the experiments to prevent cell detachment during washing steps (114). For uptake inhibition assays, cells were seeded at a density of 38,000 cells/well into PDL-coated white, clear flat-bottom 96-well plates (cat. no. CLS3903) from Corning[®] (Kennebunk, USA) one day prior to the experiments. For release assays, cells were seeded into PDL-coated ibidi µ-slides VI^{0.4} (cat. no. 80606; Gräfelting, Germany) at a density of 90,000 cells/channel.

2.3 *In vitro* uptake inhibition assay

Experiments were performed as previously described (115), with minor modifications. 100% uptake was determined in wells without addition of the SOI. To determine nonspecific substrate

uptake via other mechanisms, blank wells contained a full uptake inhibitor at a concentration that completely inhibited transporter activity. For SERT, this was 30 μM paroxetine, and for DAT and NET, 50 μM GBR12909. The signals measured from these wells were considered the baseline signal of the assay. On the day of the experiment, two seven-point logarithmic or half-logarithmic dilution series of the SOI were prepared. The first dilution series was used for pre-incubation and included only the SOI at the desired concentration in Krebs-HEPES buffer (KHB). KHB contained 10 mM HEPES (Sigma H-3375), 120 mM NaCl (Sigma S-9625), 20 mM D-(+)-glucose monohydrate (cat. no. 49159) from Sigma-Aldrich (St. Louis, MO, USA), 3 mM KCl (Merck 1.04936.500), 2 mM $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ (Merck 2382.1000), and 2 mM $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$ (Merck 5833.250). The buffer was adjusted to pH 7.3. This step allowed the compound to bind to the transporter prior to substrate addition (114). The second dilution series was the uptake solution, which contained the same concentration of SOI as in the pre-incubation solution, plus the tritiated substrate and was brought to the final volume with KHB. The substrate for SERT was 0.1 μM [^3H]-5-HT, for DAT 0.05 μM [^3H]-MPP $^+$, and for NET 0.015 μM [^3H]-MPP $^+$. DMEM was aspirated from the 96-well plate, and 200 μL of room-temperature KHB was added to each well. Each well was first treated with 50 μL of pre-incubation solution for 5 min. The wells were aspirated and refilled with 50 μL of uptake inhibition solution. The uptake time was 1 min for SERT and 3 min for DAT and NET (see Figure 7 for an overview).

Transporter	Pre-incubation time [min]	Uptake time [min]	Blocker
SERT	5	1	30 μM paroxetine
DAT	5	3	50 μM GBR 12909
NET	5	3	50 μM GBR 12909

Figure 5: Pre-incubation time, uptake time and the blockers used for the uptake inhibition assay.

The uptake inhibition solution was aspirated, and wells were washed with ice-cold KHB. After aspiration, each well was subsequently filled with 200 μL of scintillation cocktail (Ultima Gold $^{\text{TM}}$ XR). The plate was counted in a liquid scintillation counter (Wallac MicroBeta Trilux 1450-021). The value measured in the negative control was subtracted from the measured uptake values at each SOI concentration to obtain the specific uptake. Uptake in the presence of the SOI was then expressed as a percentage of a positive control (100%), which contained only the substrate without any potentially transporter-inhibiting substance. From these measured uptake values, the half-maximal concentration (IC_{50}) was determined (115). The IC_{50} value was determined by fitting a nonlinear regression curve using GraphPad Prism 8.0.1 (GraphPad Software Inc., San Diego, USA). The data were best described by a sigmoidal

dose-response model. Data were pooled from three independent experiments, each performed in triplicate.

2.4 Release assay

Experiments were conducted based on the previously described protocol (119,120), with slight modifications. This experiment was designed to test the benzofuran compounds for their release properties. Channels were pre-coated with PDL prior to seeding. 100 μL of HEK-293 cells expressing the required MAT ($0.9 \cdot 10^6$ cells/mL) were seeded into each channel of the ibidi μ -slides VI^{0.4} (cat. no. 80606; Gräfelting, Germany) and incubated overnight. On the day of the experiment, the substance solutions were prepared. Each solution was measured in triplicate per experiment, with three channels superfused with the same solution (SOI $\pm$ MON or positive control $\pm$ MON). The experiment was repeated three times. Each channel was first perfused with water for 15 min, followed by at least 5 min of KHB perfusion before insertion of the ibidi chambers containing the cells. For pre-incubation, cells were loaded with tritiated substrates: 50 μL of 0.05 μM [^3H]MPP⁺ (used for DAT), 50 μL of 0.015 μM [^3H]MPP⁺ (used for NET), and 50 μL of 0.1 μM [^3H]-5-HT (used for SERT). This loading phase was carried out for 20 min at 37 °C. Each vial was filled with 2 mL of scintillation cocktail Rotiszint®Routine (1P1A.2). Superfusion experiments were performed at concentrations around the IC₅₀ of the respective substance. Channels were superfused with SOI or positive control, both with or without MON, in KHB. The positive control consisted of 10 μM *S*-AMPH for DAT and NET, and 10 μM *p*CA for SERT. After the 20-min loading period, all channels were washed with 70 μL of room-temperature KHB. Afterwards, 90 μL were removed from each channel, and each channel was filled with 190 μL of room-temperature KHB. Superfusion was performed at a flow rate of approximately 0.5 mL/min and a targeted temperature of 21-23 °C. After insertion of the chambers into the superfusion apparatus, the system was perfused with KHB for 10 min to remove residual radiolabeled substrate and to establish a stable baseline. The experiment was then started, and fractions were collected every 2 min. From 0-6 min (3 fractions), channels were superfused with KHB to determine basal efflux, which served as the baseline. Subsequently, channels were superfused with KHB $\pm$ 10 μM MON for 8 min (4 fractions). Afterwards, all channels were superfused with their respective SOI $\pm$ MON or positive control $\pm$ MON for 10 min (5 fractions). Finally, each channel was superfused with 1% SDS solution for 6 min (3 fractions) to lyse the cells. Vials were then shaken and measured in a β -counter (Tri-Carb 2810 TR; PerkinElmer). The release in individual fractions was expressed as fractional release. First, the total radioactivity was determined by

summing all collected fractions, including those obtained by solubilization. Then, the radioactivity released in each 2-minute fraction was calculated as the percentage of the [³H]-substrate still present in the cells (= total radioactivity – radioactivity released up to that point) during the 2-min collection interval. For the calculation, Excel (Microsoft) and GraphPad Prism 8.0.1 (GraphPad Software Inc., San Diego, USA) were used. Expressed as a formula (117):

$$\text{fractional release [\%]} = \frac{\text{cpm (2 min fraction)}}{\text{total radioactivity} - \sum \text{previously released radioactivity}} \cdot 100$$

Mean ± SD were calculated from the number of experiments. Statistical significance between KHB and MON conditions at the indicated time points was assessed using a mixed-effects model with Šidák correction for multiple comparisons. Differences were considered statistically significant at $p < 0.05$.

Concentration-response curves of the SOIs were determined by applying concentrations of $0.03 \cdot \text{IC}_{50}$, $0.1 \cdot \text{IC}_{50}$, $0.3 \cdot \text{IC}_{50}$, $1 \cdot \text{IC}_{50}$, and $3 \cdot \text{IC}_{50}$. For each concentration, at least three independent experiments ($n \geq 3$) were performed, with measurements performed in triplicate. The experiment was performed as described in Chapter 2.4 (Release Assay), except that the pretreatment phase ($4 \cdot 2 \text{ min} \pm \text{MON}$) was omitted because MON was not used. Therefore, the experiment was carried out as follows: $3 \cdot 2 \text{ min}$ KHB pre-wash phase, $5 \cdot 2 \text{ min}$ SOI phase and $3 \cdot 2 \text{ min}$ 1% SDS phase. The calculations were performed in the same manner as described in Chapter 2.4. For generating the concentration-response curve, efflux was calculated after the fractional release values were obtained. This efflux was defined by averaging the release values of the SOI fractions once the release had stabilized, followed by subtracting the mean baseline value, yielding the release induced by the substance. Since each experiment was conducted in triplicate (three channels), the mean of these three channels was used for further analysis. Subsequently, the respective release values (y-axis) were plotted against their concentrations (x-axis) to determine the EC_{50} . The EC_{50} value was determined by fitting a nonlinear regression curve. The data were best described by a sigmoidal dose-response model.

3 Results

3.1 Uptake inhibition assay

First, the IC₅₀ values of the compounds were determined using an uptake inhibition assay in MAT-expressing cell lines. The results showed that the compounds in the *S*-configuration displayed significantly lower IC₅₀ values at SERT compared with their *R*-configured enantiomers. The potency of uptake inhibition varied by approximately 6-fold (4.4 μM (*S*-βK-5-MABB) to 27.4 μM (*R*-βK-5-MABB)) up to 16.5-fold (6.3 μM (*S*-βK-6-MAPB) to 104.0 μM (*R*-βK-6-MAPB)). Moreover, the data indicated that the position of the substituent on the benzofuran ring affected uptake inhibition at SERT. Benzofurans with the substituent in position 6 were approximately 2 (7.7 μM (*S*-βK-6-MABB) to 4.4 μM (*S*-βK-5-MABB)) to 3-fold (77.8 μM (*S*-βK-6-MABB) to 27.4 μM (*S*-βK-5-MABB)) less potent than those with the substituent in position 5. In contrast, the length of the alkyl side chain had only a minor impact on SERT inhibition, as the IC₅₀ values did not differ substantially (7.7 μM (*S*-βK-6-MABB) to 6.3 μM (*S*-βK-6-MAPB)).

With respect to DAT, the *S*-configured compounds also exhibited approximately 5-fold (0.115 μM (*S*-βK-6-MABB) to 0.024 μM (*R*-βK-6-MABB)) to 6-fold (0.123 μM (*S*-βK-5-MABB) to 0.708 μM (*R*-βK-5-MABB)) greater potency compared with the *R*-configured isomers. The position of the substituent on the benzofuran ring had a similar impact on potency. When the substituent was located at position 6, the compound was about 5-fold (0.024 μM (*S*-βK-6-MABB) to 0.123 μM (*S*-βK-5-MABB)) to 6-fold (0.115 μM (*R*-βK-6-MABB) to 0.708 μM (*R*-βK-5-MABB)) more potent at DAT than when the substituent was in position 5. The length of the alkyl side chain influenced potency by approximately twofold (0.024 μM (*S*-βK-6-MABB) to 0.042 μM (*S*-βK-6-MAPB) and 0.115 μM (*R*-βK-6-MABB) to 0.234 μM (*R*-βK-6-MAPB)).

For NET, stereochemistry plays a major role in determining potency, as observed for both SERT and DAT. The *S*-configured compounds exhibited approximately 1.5 (0.498 μM (*S*-βK-5-MABB) to 0.767 μM (*R*-βK-5-MABB)) to 6-fold (0.064 μM (*S*-βK-6-MABB) to 0.368 μM (*R*-βK-6-MABB)) greater potency at NET compared with the *R*-configured form. The position of the substituent also had a pronounced impact on potency. When the substituent was located at position 6, potency was a 2-fold (0.368 μM (*R*-βK-6-MABB) to 0.767 μM (*R*-βK-5-MABB)) to 8-fold (0.064 μM (*S*-βK-6-MABB) to 0.498 μM (*S*-βK-5-MABB)) higher compared with position 5. The length of the alkyl side chain influenced potency, with the 2-aminobutyl analogues being approximately 2-fold

(0.368 μ M (*R*- β K-6-MABB) to 0.750 μ M (*R*- β K-6-MAPB)) to 3-fold (0.064 μ M (*S*- β K-6-MABB) to 0.205 μ M (*S*- β K-6-MAPB)) more potent than their 2-aminopropyl counterparts.

All raw data are presented in the table below (Figure 9), which further illustrates that all tested compounds showed the strongest inhibitory effect on substrate uptake at DAT, followed by NET, while inhibition at SERT showed the lowest potency (an example: (*R*- β K-6-MABB): IC_{50} (SERT) = 77.8 μ M, IC_{50} (DAT) = 0.115 μ M and IC_{50} (NET) = 0.368 μ M).

To assess whether each compound acts as a releaser, a superfusion release assay was performed for each substance at the respective IC_{50} . An exception was made for *R*- β K-6-MABB and *R*- β K-6-MAPB, as their IC_{50} values were very high, which would require an excessive amount of compound for the superfusion release assay at the IC_{50} concentration. Therefore, for these compounds, the superfusion release assay was conducted at a concentration of 30 μ M.

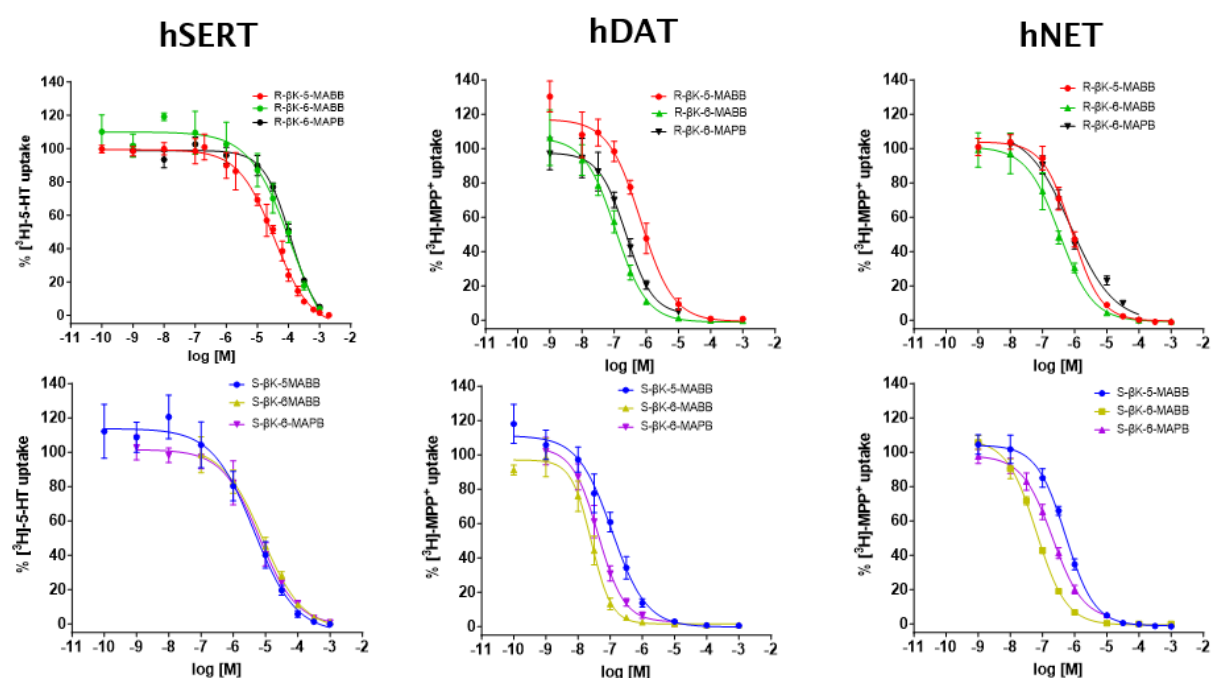


Figure 6: Representation of the uptake inhibition curves. The x-axis shows the molar concentration of the tested compounds and the y-axis indicates the percentage uptake of the radiolabeled substrate. Above the graphs, the corresponding MAT to which both curves correspond is indicated. The data points are the mean $\pm$ standard deviation (SD) from three independent experiments, each performed in triplicate. To determine the half-maximal inhibitory concentration (IC_{50}), the experimental data were plotted and fitted using non-linear regression with a sigmoidal dose-response curve.

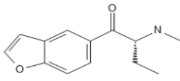
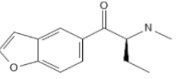
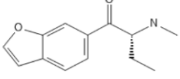
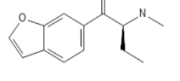
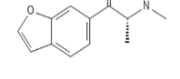
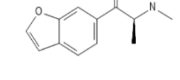
Substance	Structure	SERT [μM] $\pm$ SD	DAT [μM] $\pm$ SD	NET [μM] $\pm$ SD
<i>S</i> - β K-5-MABB		4.4 ± 1.5	0.123 ± 0.025	0.498 ± 0.078
<i>R</i> - β K-5-MABB		27.4 ± 2.6	0.708 ± 0.021	0.767 ± 0.112
<i>S</i> - β K-6-MABB		7.7 ± 0.6	0.024 ± 0.002	0.064 ± 0.006
<i>R</i> - β K-6-MABB		77.8 ± 6.7	0.115 ± 0.021	0.368 ± 0.062
<i>S</i> - β K-6-MAPB		6.3 ± 2.7	0.042 ± 0.001	0.205 ± 0.035
<i>R</i> - β K-6-MAPB		104.0 ± 13.3	0.234 ± 0.034	0.750 ± 0.112

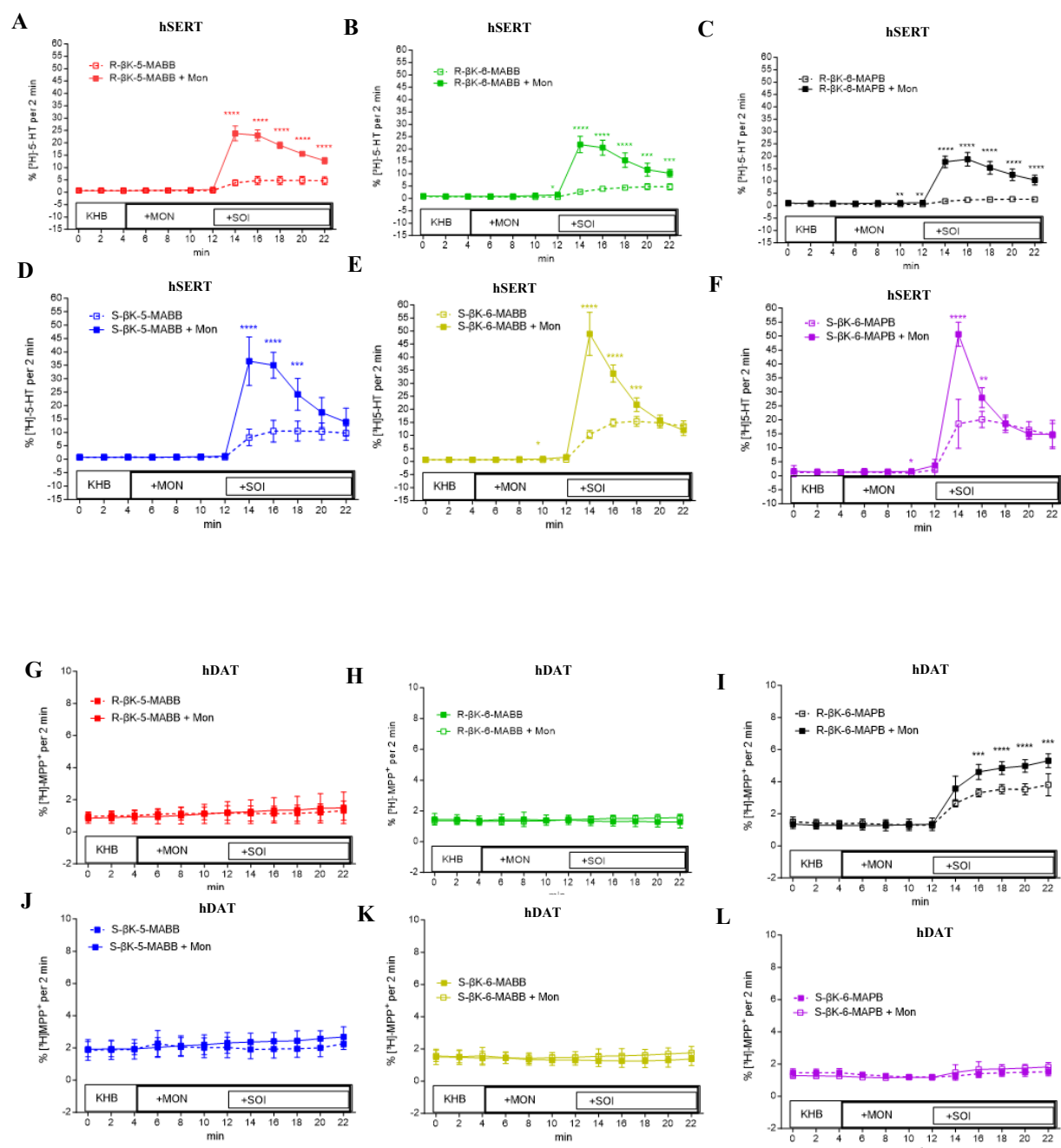
Figure 7: Overview of the results from the uptake inhibition assays. The graph shows the tested compounds, including their chemical structures and IC_{50} values for the respective MATs.

3.2 Screening for releasing activity

During the superfusion release assay, the concentrations used to determine the release properties of the compounds slightly deviated from the experimentally determined IC_{50} values (e.g., *R*- β K-5-MABB was tested at 29 μ M instead of 27.4 μ M and *S*- β K-6-MAPB at 6.5 μ M instead of 6.3 μ M). All tested compounds exhibited release at SERT, which was further enhanced by MON. The release of the compounds in the *S*-configuration was higher than that of the *R*-configuration. Moreover, the MON-induced enhancement of release was overall stronger for the *S*-configured compounds compared to the *R*-configured compounds.

At DAT, release occurred only for *R*- β K-6-MAPB, which was significantly enhanced by MON. For NET, release occurred for both *R*- β K-6-MAPB and *S*- β K-6-MAPB. The *R*-configured compound produced a higher percentage of release than the *S*-configured compound. Furthermore, the effect of MON at NET was not statistically significant. The observed difference between MON and non-MON conditions has also been reported otherwise (56).

Since all compounds induced substantial release at SERT, they were considered releasers. However, due to the very high IC_{50} values of the *R*-configured compounds, concentration-response curves were not generated for these substances, but only for the *S*-configured compounds. At DAT, *R*- β K-6-MAPB produced a pronounced increase in release. Additionally, the effect of MON was statistically significant, so this compound was also classified as a releaser. At NET, although the effect of MON was not statistically significant, the release for both compounds was sufficiently large to warrant the determination of a concentration-response curve.



(Figure continued on next page)

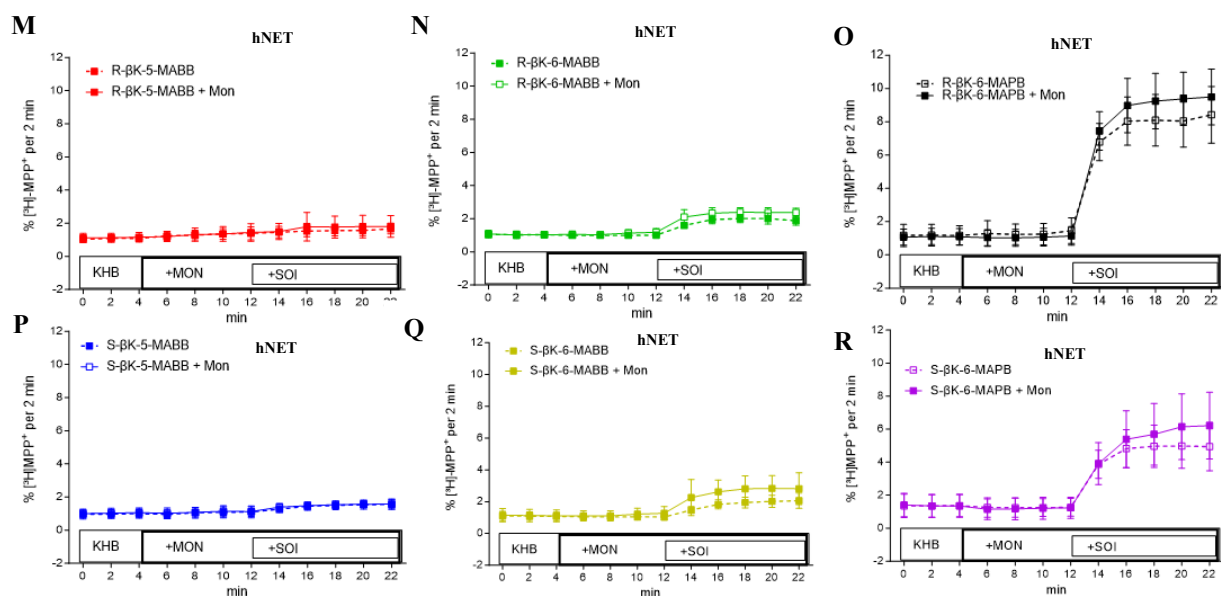


Figure 8: Release of preloaded $[^3\text{H}]\text{-5-HT}$ or $[^3\text{H}]\text{-MPP}^+$ in HEK-293 cells via stably expressing SERT, DAT or NET: Each graph shows the release profile of $[^3\text{H}]\text{-5-HT}$ at SERT (A-F), of $[^3\text{H}]\text{-MPP}^+$ at DAT (G-L) or NET (M-R) of a compound with and without MON in the superfusion release assay. The x-axis represents the time elapsed since the start of the experiment, and the y-axis displays the percentage of radiolabeled substrate released every two minutes. Compounds were applied between 14 and 24 min at concentrations near their respective IC_{50} values either in KHB (dashed line) or in the presence of $10 \mu\text{M}$ MON in KHB (solid line). Values represent the mean $\pm$ SD of 3 independent cell culture preparations ($n = 3$), performed in triplicate. Statistical significance between KHB and MON conditions at the indicated time points was assessed using a mixed-effects model with Šidák correction for multiple comparisons, confirming the releasing activity of the tested compounds. Control experiments included the known full releaser pCA for SERT and S-AMPH for DAT and NET. Differences were considered statistically significant at $p < 0.05$. Symbols indicate significance levels as follows: * $p < 0.05$, ** $p < 0.01$, *** $p < 0.001$ and **** $p < 0.0001$.

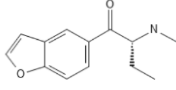
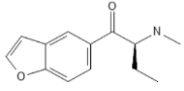
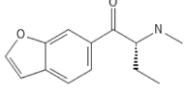
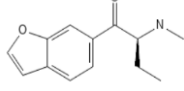
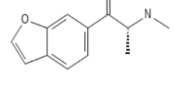
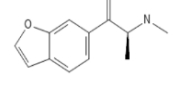
Summary of Tested Substances and Their Releasing Properties in Superfusion				
Substance	Structure	SERT	DAT	NET
<i>S</i> - β K-5-MABB		Yes	No	No
<i>R</i> - β K-5-MABB		Yes	No	No
<i>S</i> - β K-6-MABB		Yes	No	No
<i>R</i> - β K-6-MABB		Yes	No	No
<i>S</i> - β K-6-MAPB		Yes	No	Yes
<i>R</i> - β K-6-MAPB		Yes	Yes	Yes

Figure 9: Overview of the results from the superfusion release assays. The table shows the tested compounds, including their structural formulas, and indicates whether they elicit a release effect at each MAT, allowing classification as releasers.

3.3 Concentration-response analysis of releasing compounds

At SERT, all three compounds exhibited similar EC_{50} values ($EC_{50}(S\text{-}\beta K\text{-}5\text{-MABB}) = 1.07 \mu M$, $EC_{50}(S\text{-}\beta K\text{-}6\text{-MABB}) = 1.95 \mu M$ and $EC_{50}(S\text{-}\beta K\text{-}5\text{-MAPB}) = 1.13 \mu M$). Regarding maximal release (E_{max}), $S\text{-}\beta K\text{-}6\text{-MAPB}$ induced approximately twice the release of $S\text{-}\beta K\text{-}5\text{-MABB}$ ($E_{max}(S\text{-}\beta K\text{-}6\text{-MAPB}) = 15.25 \%$ to $E_{max}(S\text{-}\beta K\text{-}5\text{-MABB}) = 7.36 \%$) and about 1.2-fold the release of $S\text{-}\beta K\text{-}6\text{-MABB}$ ($E_{max}(S\text{-}\beta K\text{-}6\text{-MAPB}) = 15.25 \%$ to $E_{max}(S\text{-}\beta K\text{-}6\text{-MABB}) = 12.61 \%$). These values were normalized to pCA , indicating that $S\text{-}\beta K\text{-}5\text{-MABB}$ corresponded to approximately 41.6 %, $S\text{-}\beta K\text{-}6\text{-MABB}$ approximately 71.2 %, and $S\text{-}\beta K\text{-}6\text{-MAPB}$ approximately 86.1 % of pCA 's release ($E_{max}(pCA) = 17.71 \%$).

At DAT, $R\text{-}\beta K\text{-}6\text{-MAPB}$ had an EC_{50} of $0.105 \mu M$ and an E_{max} of 3.13 %. Compared to $S\text{-AMPH}$, $R\text{-}\beta K\text{-}6\text{-MAPB}$ produced approximately 33.1 % of $S\text{-AMPH}$'s release.

At NET, $S\text{-}\beta K\text{-}6\text{-MAPB}$ and $R\text{-}\beta K\text{-}6\text{-MAPB}$ both exhibited nearly identical EC_{50} values ($EC_{50}(S\text{-}\beta K\text{-}6\text{-MAPB}) = 0.115 \mu M$ and $EC_{50}(R\text{-}\beta K\text{-}6\text{-MAPB}) = 0.113 \mu M$). In terms of E_{max} , $R\text{-}\beta K\text{-}6\text{-MAPB}$ induced 1.3-fold higher maximal release than $S\text{-}\beta K\text{-}6\text{-MAPB}$ ($E_{max}(R\text{-}\beta K\text{-}6\text{-MAPB}) = 7.88 \%$ to $E_{max}(S\text{-}\beta K\text{-}6\text{-MAPB}) = 6.01 \%$). When compared to $S\text{-AMPH}$, $S\text{-}\beta K\text{-}6\text{-MAPB}$ showed similar maximal release to $S\text{-AMPH}$, whereas $R\text{-}\beta K\text{-}6\text{-MAPB}$ exhibited 1.3-fold higher maximal release than $S\text{-AMPH}$ ($E_{max}(S\text{-AMPH}) = 6.05 \%$).

$R\text{-}\beta K\text{-}6\text{-MAPB}$ induced release at all three MATs, with NET showing approximately 2.5-fold higher release ($E_{max}(R\text{-}\beta K\text{-}6\text{-MAPB}, NET) = 7.88 \%$ to $E_{max}(S\text{-}\beta K\text{-}6\text{-MAPB}, DAT) = 3.13 \%$). $S\text{-}\beta K\text{-}6\text{-MAPB}$ produced release at both SERT and NET, with the effect at SERT being about 2.5-fold stronger ($E_{max}(S\text{-}\beta K\text{-}6\text{-MAPB}, SERT) = 15.25 \%$ to $E_{max}(S\text{-}\beta K\text{-}6\text{-MAPB}, NET) = 6.01 \%$). In contrast, $S\text{-}\beta K\text{-}5\text{-MABB}$ and $S\text{-}\beta K\text{-}6\text{-MABB}$ both elicited release exclusively at SERT.

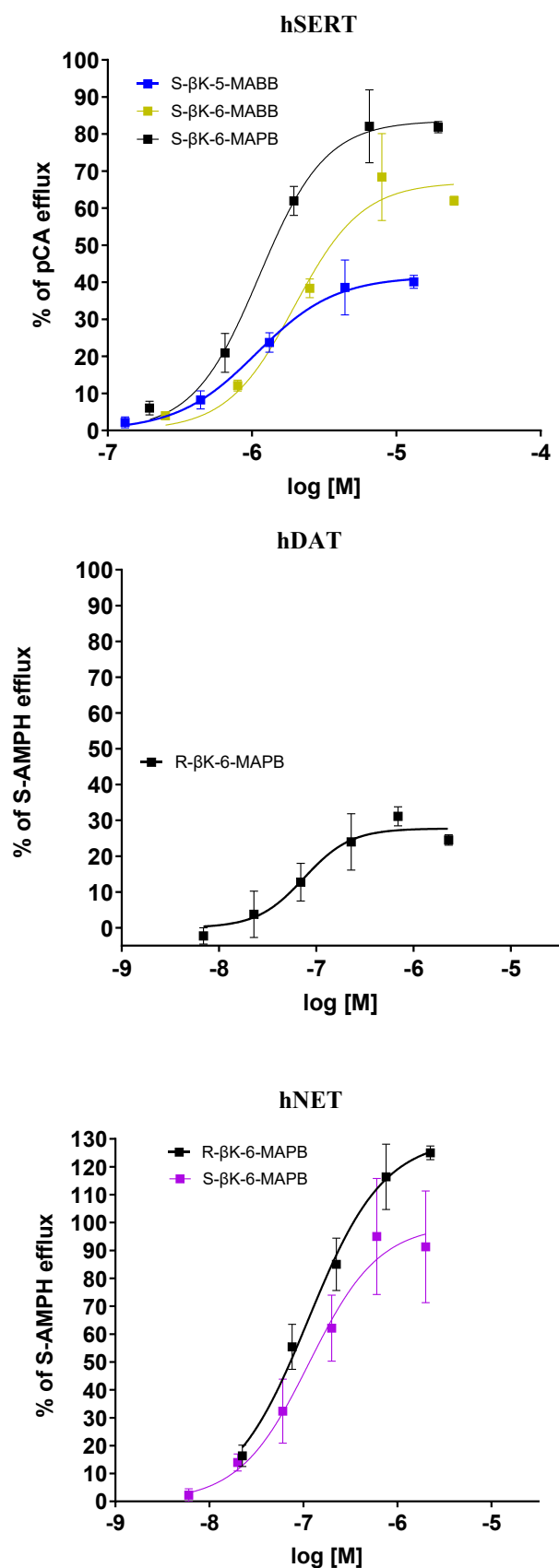


Figure 10: Release of preloaded [^3H]-substrate in HEK-293 cells stably expressing MAT. Each graph shows the release profile of [^3H]-substrate for a single compound in the superfusion release assay at one MAT. The x-axis represents the logarithmic molar concentration of the compound, while the y-axis displays the percentage of radiolabeled substrate released relative to the control substance. Values represent the mean $\pm$ SD of three

independent cell culture preparations ($n = 3$), each performed in triplicate. Control experiments included the known full releaser for the respective MAT.

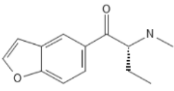
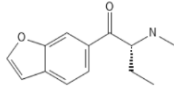
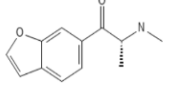
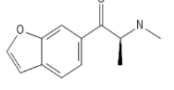
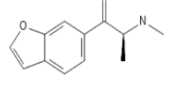
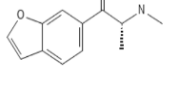
MAT	Substance	Structure	EC ₅₀ (95 % CI) [μM]	E _{max} (95 % CI) [%]	% (of full releaser)
SERT	<i>S</i> -βK-5-MABB		1.071 (0.784 – 1.463)	7.36 (6.49 – 8.24)	41.6
SERT	<i>S</i> -βK-6-MABB		1.947 (1.444 – 2.625)	12.61 (11.09 – 14.13)	71.2
SERT	<i>S</i> -βK-6-MAPB		1.127 (0.924 – 1.373)	15.25 (14.24 – 16.26)	86.1
DAT	<i>R</i> -βK-6-MAPB		0.105 (0.043 – 0.260)	3.13 (1.91 – 4.34)	33.1
NET	<i>R</i> -βK-6-MAPB		0.113 (0.079 – 0.160)	7.88 (7.02 – 8.74)	130.2
NET	<i>S</i> -βK-6-MAPB		0.115 (0.065 – 0.204)	6.01 (4.81 – 7.21)	99.3

Figure 11: Concentration-response curve results from the superfusion release assay, showing each tested compound with its corresponding transporter, potency (EC_{50}), efficacy (E_{max}) and release percentage relative to a standard full releaser (pCA for SERT and S-AMPH for DAT and NET).

4 Discussion

Amphetamines play an important role in neuropsychopharmacology and have both therapeutic and abuse-related properties. Recent phase 3 studies indicate increasing interest in MDMA, which could serve as a therapeutic agent in the treatment of PTSD (107,108) and have led to granting market access for MDMA in Australia (121) and off-label use in Switzerland (122,123). One goal of current research efforts in the pharmacotherapeutic treatment of PTSD is to develop MDMA derivatives that exhibit improved pharmacological characteristics, such as an enhanced safety profile and potentially higher therapeutic efficacy (56). In this context, benzofuran derivatives may represent promising therapeutic agents. The therapeutic effect of MDMA is thought to primarily rely on transporter-mediated release of serotonin (124), although additional mechanisms such as oxytocin (64) may contribute. A disadvantage of amphetamines is their noradrenergic activity, which can influence the cardiovascular system and may lead to increased contractile force, elevated heart rate, and cardiovascular strain (125). A reduced effect on the noradrenergic system may therefore mitigate these cardiotoxic effects and may be associated with an improved safety profile. Since the addictive effects of psychostimulants are associated with the dopaminergic system (9), reduced transporter-mediated release or influence on the dopaminergic system via DAT may reduce dopaminergic reinforcement mechanisms associated with abuse liability. Therefore, release at DAT and NET may be undesirable for the use of benzofurans as alternatives to MDMA, whereas release at SERT may prove to be beneficial.

A suitable comparison to existing data is provided by a previous master's thesis conducted in the Sitte laboratory, which analysed both stereoisomers of β K-5-MAPB (126). Although these data are not peer-reviewed, they have the advantage that the experiments were conducted in the same laboratory under identical experimental conditions, allowing better comparability. Data for MDMA are available under similar experimental conditions, reporting IC₅₀ values and release profiles at MATs (56).

Regarding uptake inhibition, the present findings are consistent with previous observations showing that benzofuran cathinones exhibit the highest inhibitory potency at DAT, followed by NET, and SERT (126). Compared with MDMA (56), the benzofuran cathinones tested in the current study exhibited stronger inhibitory potency at all three MATs (see Figure 7), suggesting that structural modifications, including the β -keto group and benzofuran ring system, may contribute to enhanced transporter inhibition. These findings further allow an assessment of the structural determinants of MAT inhibition.

Among the investigated structural features, stereochemistry had the strongest and most consistent influence on inhibitory potency across SERT, DAT, and NET. The *S*-enantiomer exhibited higher potency than the *R*-enantiomer, suggesting a stereoselective interaction with the binding sites of the transporters. This is consistent with previous findings (124,126), although the *R*-enantiomer was reported to be more potent at DAT than the *S*-enantiomer (124). Despite being performed in different experimental systems (HEK-293 cells vs. rat brain synaptosomes), these studies demonstrated a substantial difference in potency between stereoisomers. Taken together, these findings suggest that the *R*-enantiomers contribute less strongly to transporter inhibition.

The position of the substituent on the benzofuran ring further modulates uptake inhibition. At SERT, substitution at position 6 moderately reduced inhibition, while at DAT and NET substitution at position 6 enhanced potency. This observation aligns with previous findings (41,124,127). However, one study reported slightly higher potency at NET for substitution in position 6 (41). These findings further support the hypothesis (41) that an oxygen atom in the para position relative to the aminoalkyl side chain enhances serotonergic properties while reducing dopaminergic ones.

The length of the alkyl chain also influences the potency of the compound: β K-6-MABB, which contains a butyl chain, exhibited stronger inhibition than its corresponding propyl analogue β K-6-MAPB across all tested transporters, except for the *S*-enantiomer at SERT. This deviation from the general trend may be due to the high variability of the data, as reflected by overlapping standard deviations. A previous study reported increased potency at SERT and DAT for the butyl analogues but observed the opposite effect at NET (42). However, these discrepancies should be interpreted with caution and may arise from differences in steric and conformational properties of the respective ring systems, which could influence how the alkyl side chains interact with the transporter binding site.

Taken together, these findings indicate that the stereochemistry of the compounds strongly influences inhibitory potency at all three MATs, with the *S*-form representing the more potent enantiomer. In addition, both the position of the substituent on the benzofuran ring and the length of the alkyl chain affect inhibitory activity. Para-substitution appears to enhance serotonergic effects while reducing dopaminergic effects, and the butyl chain generally confers stronger inhibition than the propyl chain. These effects are particularly relevant for compounds with steep concentration-response curves, where inhibition increases sharply within a narrow concentration range, as observed for the tested compounds.

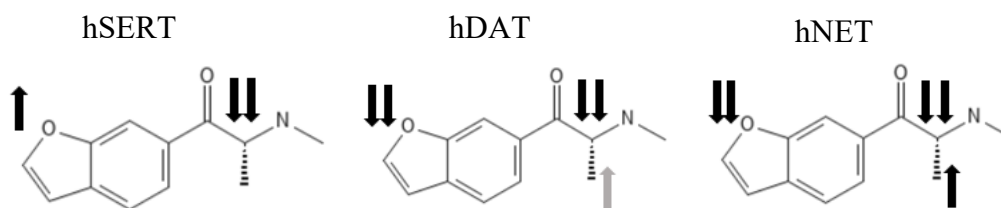


Figure 12: Uptake inhibition effects of benzofuran cathinones at MATs. An upward arrow indicates an increase in IC_{50} , whereas a downward arrow indicates a decrease. The number of arrows reflects the magnitude of the effect: more arrows correspond to a stronger influence. The effect of stereochemistry is indicated above the chiral carbon. A gray arrow indicates an uncertain effect, while black arrows indicate more likely effects.

Since uptake inhibition assays do not distinguish between transporter substrates and non-transported competitive inhibitors (116,117), the superfusion release assay was necessary to further characterize releaser properties.

With respect to releasing properties, the data indicated that all tested substances acted as releasers at SERT, as all compounds showed a significant enhancement in the presence of MON. At DAT and NET, only compounds with a propyl chain showed release, whereas the remaining compounds showed no detectable effect. The release at DAT was restricted to the *R*-enantiomer of the propyl form. At NET, the propyl derivatives showed release, but MON did not significantly enhance this effect. Nevertheless, because the increase in release was pronounced and the substrate MPP^+ cannot passively diffuse into the cell due to its charge (128), the observed effect strongly suggests transporter-mediated release rather than uptake inhibition. The literature presents inconsistent findings regarding transporter-specific release. Studies conducted in rat brain synaptosomes (124,129) as well as in HEK-293 cells (126,127) do not show a uniform pattern across transporters. However, unlike MDMA (56), which induces release at SERT, DAT, and NET, *S*- β K-5-MABB and *S*- β K-6-MABB only showed release at SERT, suggesting a more selective release profile. The release data further allow an assessment of how structural modifications influence transporter-mediated monoamine release.

At SERT, stereochemistry appears to strongly influence release properties, as the *S*-form produced stronger release than the *R*-form at the IC_{50} , both in the presence and absence of MON. This is consistent with findings from the previous master thesis mentioned before (126). This interpretation assumes that the release of the *R*-form does not increase disproportionately at concentrations outside the IC_{50} range. All three tested *S*-enantiomers exhibited EC_{50} values between 1-2 μ M, indicating that they act within a very similar concentration range and are therefore similarly potent in inducing efflux. Similar EC_{50} values for benzofurans were reported within a range of 0.2-3.9 μ M (126). A key difference of the tested benzofurans, however, lies in their efficacy. The position of the substituent on the benzofuran ring seems to have a strong influence on efficacy, ranging from *S*- β K-5-MABB ($E_{max} = 7.4\%$) to *S*- β K-6-MABB

($E_{\max} = 12.6\%$). In addition, the alkyl-chain length appears to influence efficacy, with the 2-aminopropyl group producing a stronger effect ($E_{\max} = 15.3\%$) than the 2-aminobutyl group ($E_{\max} = 12.6\%$). Compared with the full releaser *p*CA (17.7 %), *S*- β K-5-MABB exhibited only about 42 % of *p*CA's release, which may indicate partial releasing properties. For *S*- β K-6-MABB, this value was about 71 % *p*CA's release, indicating stronger releasing properties, whereas *S*- β K-6-MAPB reached 86 % *p*CA's release and appears to act as a full releaser. In summary, EC_{50} values varied within a relatively narrow micromolar range, while $E_{\max}$ values ranged from partial to near-full release. This suggests that structural differences primarily influence intrinsic efficacy rather than functional potency within the tested range. These findings are supported by mechanistic models of MATs, which show that partial releasers exhibit reduced maximal release efficacy compared with full releasers (130). Possible explanations for these observed variations in $E_{\max}$ may arise from ligand-specific conformational dynamics of the transporter, as different ligands stabilize specific transporter states that could influence maximal release (131).

The concentration-response curve at DAT shows that *R*- β K-6-MAPB produced moderate release, with an $E_{\max}$ of approximately 33 %, relative to *S*-AMPH. Therefore, it may be classified as a partial releaser. Such partial release effects at DAT have also been reported in the literature (132). One possible explanation is that different DAT ligands may stabilize distinct conformational states of the transporter (133). Consistent with these findings, DAT-mediated release was observed only for the *R*-enantiomer of the benzofuran cathinone, whereas different results were reported for the corresponding amphetamine analogue (126).

At NET, the data showed that the length of the alkyl chain was a key determinant of release, as only the 2-aminopropyl group produced release, whereas no release was observed for the 2-aminobutyl group. The available literature reports varying findings regarding NET release. While some studies report NET release for propyl derivatives (126,127,129), another study observed release also with butyl-substituted compounds (124). It should be noted, however, that the latter study used a different experimental system (rat brain synaptosomes) and only tested amphetamine analogues, which could explain the discrepancy. Another observation is that the *R*-enantiomer (7.9 %) produced a stronger release effect than the *S*-enantiomer (6.0 %), even though the 95 % confidence intervals slightly overlapped (*R*- β K-6-MAPB = 7.88 % (7.02 % - 8.74 %) and *S*- β K-6-MAPB = 6.01 % (4.81 % - 7.21 %)). This finding is consistent with the literature (126). However, it contrasts with results obtained for the amphetamine derivatives. Consistent with these observations, another study (124) also reported that the *R*-enantiomers induced lower maximal release.

R-βK-6-MAPB acts as a full releaser at NET but only as a partial releaser at DAT. Such transporter-specific differences have also been reported for mephedrone (132). These differences may arise from dynamic, conformational, or structural differences between the transporters. Studies conducted in rat brain synaptosomes (124,129) as well as in HEK-293 cells (126,127) do not show a uniform pattern across the tested transporters. When considering all available data, a trend may possibly be observed, suggesting that SERT and NET tend to exhibit higher maximal release ($E_{\max}$) compared to DAT. However, a study performed in rat synaptosomes (124) reported no significant differences between transporters, whereas a study in HEK-293 cells observed higher release at hDAT than at hSERT for a benzofuran cathinone analogue (126). Therefore, transporter-specific differences in $E_{\max}$ appear to be context-dependent and influenced by the experimental system. In the present study, release at NET and DAT was also observed exclusively for propyl-substituted compounds, whereas butyl analogues failed to induce release at these transporters. This may indicate that alkyl chain length contributes to transporter selectivity in release assays. Furthermore, the corresponding amphetamine analogues exhibited stronger releasing properties overall (126). Taken together, the present findings indicate that structural modifications in benzofuran cathinones modulate transporter selectivity as well as maximal release efficacy in a context-dependent manner. For the concentration-response experiments used to characterize release properties, temperature was checked and documented prior to each experiment to ensure valid measurements.

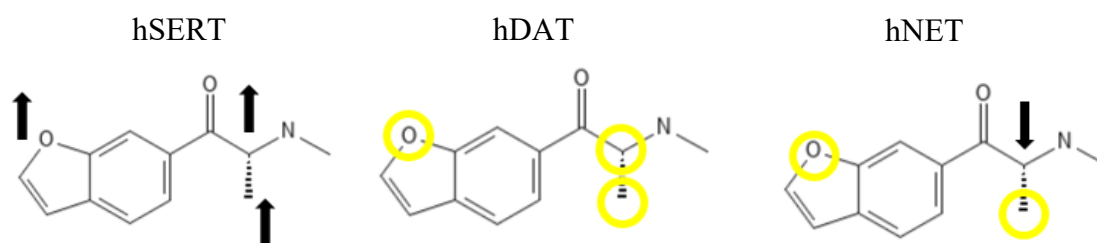


Figure 13: Effects of structural features of benzofuran cathinones on their release efficacy ($E_{\max}$) at MATs. The effect of stereochemistry is indicated above the chiral carbon atom. Yellow-circled areas highlight the key structural features necessary to elicit a release effect at MATs.

5 **Conclusion**

Based on the previously discussed findings, *S*- β K-6-MAPB may be less suitable as an alternative substance for treating PTSD. In contrast, *S*- β K-5-MABB and *S*- β K-6-MABB demonstrate SERT-mediated release without significant release activity at DAT or NET, suggesting a more targeted and potentially safer pharmacological profile. Their strong uptake inhibition of radiolabeled substrates at DAT and NET, without inducing release, suggests direct transporter inhibition rather than substrate-mediated release. As transporter substrates are generally expected to induce release according to the alternating access model, the absence of release supports a sole inhibitory mechanism. The extent of SERT-mediated monoamine release required for therapeutic effects in PTSD has not yet been clearly established. For a more precise classification, investigations using patch-clamp techniques and *in vivo* examinations utilizing behavioral pharmacological experiments would be needed. Overall, these findings suggest that selective serotonergic release can be achieved through structural modifications in benzofuran cathinones. This may improve their pharmacological profile for the treatment of PTSD, provided that future toxicological studies confirm their safety.

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7 Appendix

7.1 Zusammenfassung

Neue psychoaktive Substanzen gewinnen zunehmend an Bedeutung im Bereich der öffentlichen Gesundheit und könnten neue therapeutische Ansätze für psychiatrische und kognitive Störungen ermöglichen. MDMA wird derzeit in Phase-3-Studien als Ergänzung zur Psychotherapie zur Behandlung von PTBS untersucht. Nach zwei Phase-3-Studien forderte die FDA in einem *Complete Response Letter* weitere Untersuchungen zur Sicherheit und Wirksamkeit. Dies führte zur Entwicklung von MDMA-Analoga mit potenziell verbesserten pharmakologischen Profilen.

In dieser Masterarbeit wurden sechs Benzofurancathinon-Derivate *in vitro* hinsichtlich ihrer Aktivität an den Monoamintransportern SERT, DAT und NET untersucht. Hierzu wurden Wiederaufnahmehemmungs- und superfusionsbasierte Freisetzungssassays in HEK-293-Zellen durchgeführt und IC₅₀-Werte sowie Freisetzungseigenschaften bestimmt.

Die Ergebnisse zeigten, dass Stereochemie, Substituentenposition am Benzofuranring und Alkylkettenlänge die IC₅₀ Werte beeinflussen. Die Potenz variierte zwischen den Transportern mit der höchsten Potenz am DAT, gefolgt von NET und SERT. Alle untersuchten Substanzen induzierten Freisetzung an SERT. Am DAT wurde Freisetzung nur für *R*-βK-6-MAPB beobachtet, während am NET beide Stereoisomere von βK-6-MAPB Freisetzung zeigten.

Diese Ergebnisse deuten darauf hin, dass ausgewählte Benzofurancathinone eine selektive serotonerge Aktivität aufweisen könnten und somit potenziell vielversprechende Kandidaten für eine weitere pharmakologische Evaluierung in der Behandlung von PTBS darstellen.